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# Intracellular $\text{Ca}^{2+}$ release by $\text{InsP}_3$ in plants and effect of buffers on $\text{Ca}^{2+}$ diffusion

JOEL ALEXANDRE AND JEAN-PAUL LASSALLES

Laboratoire 'Echanges Cellulaires', URA CNRS 203, Faculté des Sciences, BP 118, 76134 Mont-Saint-Aignan, Cedex, France

## SUMMARY

Several evidences favour the idea that  $\text{InsP}_3$ , a well known second messenger in animal cells, can also be used by plants. The studies supporting this hypothesis are reviewed, with special emphasis on the experiments concerning the release of vacuolar  $\text{Ca}^{2+}$  by  $\text{InsP}_3$ . We also report patch-clamp experiments on isolated red beet vacuoles. In some of the patches, the  $\text{InsP}_3$ -mediated  $\text{Ca}^{2+}$  channel exhibited a 50 pS conductance state beside the 30 pS state mentioned in a previous work. The  $\text{InsP}_3$ -mediated vacuolar current was smaller with  $\text{Ba}^{2+}$  or  $\text{Sr}^{2+}$  ions replacing  $\text{Ca}^{2+}$  in both the pipette and external solutions. Heparin ( $1 \text{ mg ml}^{-1}$ ) in the extravacuolar medium decreased by 60% only, the  $\text{InsP}_3$ -mediated  $\text{Ca}^{2+}$  current. The inhibitory effect of heparin was not specific: with the same concentration used, another current in the tonoplast was reduced by 46%. The coupling by diffusion of  $\text{Ca}^{2+}$  ions between the  $\text{InsP}_3$ -mediated  $\text{Ca}^{2+}$  channel and the  $\text{Ca}^{2+}$ -sensitive channel in the tonoplast has been studied. The coupling was dependent on the  $\text{Ca}^{2+}$  buffers (BAPTA or EGTA) in the extravacuolar medium. At least 20 mM BAPTA were necessary to prevent coupling between channels. In the range of concentrations tested (less than 30 mM), EGTA could not prevent coupling. A review of the theoretical models describing diffusion in a buffered medium is presented, and a simplified version from the models is proposed to discuss our results.

## 1. INTRODUCTION

In a few cases, variations in cytosolic  $\text{Ca}^{2+}$  have been reported to accompany the transduction of signals in plant cells (Gehring *et al.* 1990; McAinsh *et al.* 1990; Schroeder & Hagiwara 1990; Gilroy *et al.* 1991; Knight *et al.* 1991). There are two basic questions concerning the occurrence of a  $\text{Ca}^{2+}$  change in the cytosol. The first one deals with the origin of the signal: the ions used to modify  $\text{Ca}^{2+}$  concentration can be provided by extracellular (Schroeder & Hagiwara 1990) or (and) intracellular (Gilroy *et al.* 1991)  $\text{Ca}^{2+}$  pools. One of the striking differences between animals and plants is the presence of a large vacuolar  $\text{Ca}^{2+}$  reservoir in plant cells. At least in principle, the less than  $\mu\text{M}$  cytosolic  $\text{Ca}^{2+}$  could be modified by  $\text{Ca}^{2+}$  from the vacuolar compartment only. It is now established that  $\text{InsP}_3$  can mobilize this  $\text{Ca}^{2+}$  pool by opening a channel in the tonoplast. In this paper, we review previous studies on the possible use of  $\text{InsP}_3$  by plant cells, and we present our last results on the  $\text{InsP}_3$ -mediated channel from isolated red beet vacuoles.

A second question concerns the propagation of the  $\text{Ca}^{2+}$  signal itself. Even if it is still common use to speak of 'the' cytosolic  $\text{Ca}^{2+}$  concentration, it must be kept in mind that the distribution of  $\text{Ca}^{2+}$  ions may be non uniform in the cytosol. Recent measurements using  $\text{Ca}^{2+}$  imaging by fluorescent dyes, describe spatial and temporal variations in cytosolic  $\text{Ca}^{2+}$

during the transduction of a cellular signal (Gilroy *et al.* 1991). When a channel or a  $\text{Ca}^{2+}$  pump are activated in a membrane delimiting the cytosolic compartment, the ions released near these active sites will eventually spread in this compartment. The propagation of this  $\text{Ca}^{2+}$  signal depends on the structural properties of the cytosol. There are some evidences for the influence of such a structure in the propagation of 'wave like' signals. However, very near the sites (channels or pumps), it is usually assumed that the movement of  $\text{Ca}^{2+}$  ions is only due to diffusion: the  $\text{Ca}^{2+}$  buffer strength of the cytosol is the only 'structural' parameter in this case. We will review the theoretical models used to explain the effect of buffers on the propagation of a  $\text{Ca}^{2+}$  signal near channels or pumps.

In some experiments, buffers are used to disrupt a stimulus-response coupling. When there is an extended chain of reactions between the stimulus and the response, the buffer is expected to block the step(s) involving a change in cytosolic  $\text{Ca}^{2+}$ . The isolated red beet vacuole offers a very simple example of coupling between a stimulus and its response at the cellular level. The stimulus ( $\text{InsP}_3$ ) opens a  $\text{Ca}^{2+}$  channel in the tonoplast, and the  $\text{Ca}^{2+}$  signal is represented by the diffusion of  $\text{Ca}^{2+}$  ions in the extravacuolar medium. The response is the change in activity of a  $\text{Ca}^{2+}$ -sensitive channel, also present in the tonoplast. We report here patch-clamp experiments performed on red beet vacuoles to study the effects of buffers on a

Ca<sup>2+</sup> signal. We analyse these effects by using a simplified model for the diffusion of Ca<sup>2+</sup> in a buffered medium.

## 2. THE USE OF INSP<sub>3</sub> BY PLANT CELLS

### (a) *Phosphoinositides and inositol phosphates in plants*

Reviews covering the different aspects of phosphoinositide metabolism in plants have been recently published (Boss 1988, Lehle 1990). They insist upon technical difficulties in the identification and quantification of inositol phosphates in different plant tissues. In fusogenic carrot cells, Rincon *et al.* 1988 find no evidence for the existence of Ins(1, 4)P<sub>2</sub> or InsP<sub>3</sub>. In contrast, in *Samanea saman* pulvini, a short light treatment elicits a rapid change in phosphoinositide components (Morse *et al.* 1987) and a 30% increase in Ins(1, 4)P<sub>2</sub> plus InsP<sub>3</sub> is detected in this tissue. Since the first report on the existence of phosphoinositides in plants (Boss & Massel 1985), intensive studies have established that plant and animal cells do contain polyphosphoinositol lipids.

### (b) *Enzymes system*

One of the interesting regulations by Ca<sup>2+</sup> ions, involving enzymes in the phosphatidylinositol cycle has been found in plants for physiological concentrations of Ca<sup>2+</sup> (Kamada & Muto 1991). Important differences in the regulation of phosphoinositide metabolism have been established between plants and animals. The phosphatidylinositol bisphosphate level is much lower in plants (Coté *et al.* 1989) than in animal cells (Abdel-Latif *et al.* 1985). A report by Drobak *et al.* (1991) demonstrates that enzymes involved in inositol phosphate metabolism are present in plant tissues. The authors have detected configurations of inositol phosphates which are not mentioned in other mammalian systems. This suggests that metabolism of InsP<sub>3</sub> is complex and very particular, when compared to other eukaryote cells.

### (c) *Site(s) of action of InsP<sub>3</sub>*

The InsP<sub>3</sub>-mediated Ca<sup>2+</sup> release has been studied according to various experimental approaches. Most of the protocols use exogenous InsP<sub>3</sub>. The first evidence for an InsP<sub>3</sub>-mediated Ca<sup>2+</sup> release has been obtained from experiments performed on microsomes from zucchini (Drobak & Ferguson 1985). Other authors also working on subcellular fractions, have confirmed this result (Reddy & Poovaiah 1987; Canut *et al.* 1989; Lehle & Ettlinger 1990). Ca<sup>2+</sup> is first loaded into intracellular compartments in the presence of ATP. After accumulation of Ca<sup>2+</sup> by an ATP-dependent Ca<sup>2+</sup> pump located in the microsome, the release of Ca<sup>2+</sup> following addition of InsP<sub>3</sub> is measured. The release is transient and rapid (less than 0.5 min); it is followed by a slower period of Ca<sup>2+</sup> reuptake. In these experiments, the amount of Ca<sup>2+</sup> released never exceeds 20% of the Ca<sup>2+</sup> sequestered

into the microsomes (Brosnan 1990). Monovalent ions seem to be required to observe InsP<sub>3</sub>-mediated Ca<sup>2+</sup> release (Canut *et al.* 1989). The data presented suggest that an influx of potassium functions to balance the efflux of positive charges during the Ca<sup>2+</sup> release. In most of the experiments with microsomes, membrane fractions were not defined with enough accuracy, and it was not possible to determine which of the cell membranes contained the InsP<sub>3</sub> receptor. Recent studies demonstrate that at least *in vitro*, the vacuole responds to InsP<sub>3</sub> by releasing Ca<sup>2+</sup> in the external medium. Schumaker & Sze (1987) use tonoplast vesicles from oat roots, preloaded with calcium via a H<sup>+</sup>/Ca<sup>2+</sup> antiport system. They demonstrate that InsP<sub>3</sub> specifically releases Ca<sup>2+</sup> in a concentration dependent manner. Ranjeva *et al.* (1988) confirm the previous data, by monitoring the fluorescence of the Ca<sup>2+</sup> released in the external medium by intact vacuoles isolated from *Acer pseudoplatanus*. Cornelius *et al.* (1989) also measured a Ca<sup>2+</sup> release from intact vacuoles (*Neurospora crassa*) with two independent methods (calcium binding to fura-2 and release of <sup>45</sup>Ca). In summary, many reports indicate that there is an InsP<sub>3</sub>-sensitive site in the vacuolar membrane. However, one cannot discard other sites on intracellular membranes like the endoplasmic reticulum.

### (d) *Mechanism of the InsP<sub>3</sub>-mediated Ca<sup>2+</sup> release*

The data previously reported have strongly suggested that the release of Ca<sup>2+</sup> by InsP<sub>3</sub> occurs through an ionic channel. Using patch-clamp techniques (Hamill *et al.* 1981) on isolated red beet vacuoles, Alexandre *et al.* (1990) demonstrate that InsP<sub>3</sub> specifically releases Ca<sup>2+</sup> by activating a Ca<sup>2+</sup> channel. This channel is activated by voltage over the physiological range (−20 to +100 mV, with potential referred to the cytosol). The single channel conductance is 30 pS (5 mM Ca<sup>2+</sup> on the vacuolar side and 1 mM on the cytoplasmic side). A higher conductance state was sometimes recorded in both vacuole-attached and outside out configurations. In most patches, many flickering events were present for both states. Figure 1 describes an experiment performed in a vacuole-attached patch. Similar records (*n* = 5) with substates give values of 30 pS and 50 pS for the two conductance states. The presence of multiple conductance states is found with InsP<sub>3</sub>-gated channels in the reticulum of canine cerebellum. The channel exhibits four conductance levels that are multiples of a unit conductance step. The vacuolar current induced by 10<sup>−6</sup> M InsP<sub>3</sub> was determined as explained in a previous report (Alexandre *et al.* 1990). The other currents were blocked with 10<sup>−3</sup> M Ca<sup>2+</sup> and 10<sup>−4</sup> M Zn<sup>2+</sup> (Hedrich & Neher 1987, Hedrich & Kurkdjian 1988). Figures 2 and 3 describe experiments performed in whole-vacuole configuration. As shown in figure 2, with mole-for-mole replacement of CaCl<sub>2</sub> with BaCl<sub>2</sub> or SrCl<sub>2</sub> in both external and pipette solutions, the InsP<sub>3</sub>-induced currents were smaller for Ba<sup>2+</sup> and Sr<sup>2+</sup> than for Ca<sup>2+</sup> ions. In all cases, these currents increased for positive potentials and were blocked for potentials more negative than −20 mV.

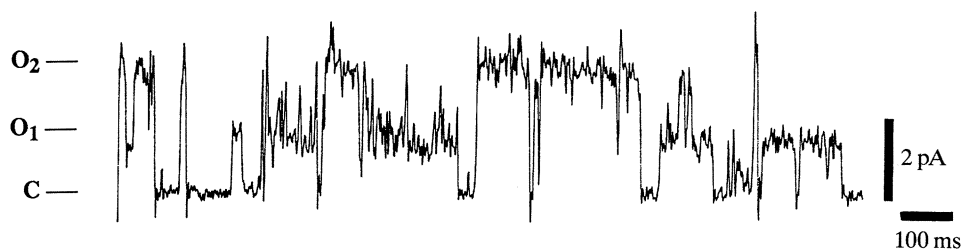


Figure 1. Conductance states in the  $\text{InsP}_3$ -activated  $\text{Ca}^{2+}$  channel. Single-channel recording for a vacuole attached patch with  $10^{-6}$  M  $\text{InsP}_3$ ,  $10^{-3}$  M  $\text{CaCl}_2$ ,  $10^{-4}$  M  $\text{ZnCl}_2$  in the pipette and external solutions ( $5 \times 10^{-3}$  M Tris, pH adjusted to 7.5 with MES). Pipette potential:  $-60$  mV.  $\text{O}_1$ ,  $\text{O}_2$ : conductance states; C: closed state. The large conductance state was obtained in outside-out patches as well, but it was not present in each of the patches tested. For all the experiments described, figure 1 to figure 5, the pipette and the external solutions also contained: 0.1 M KCl, 0.6 M Sorbitol,  $2 \times 10^{-3}$  M  $\text{MgCl}_2$ . The preparation of vacuoles, the current recordings and the analysis of data have been described elsewhere (Alexandre *et al.* 1990; Alexandre & Lassalles 1991).

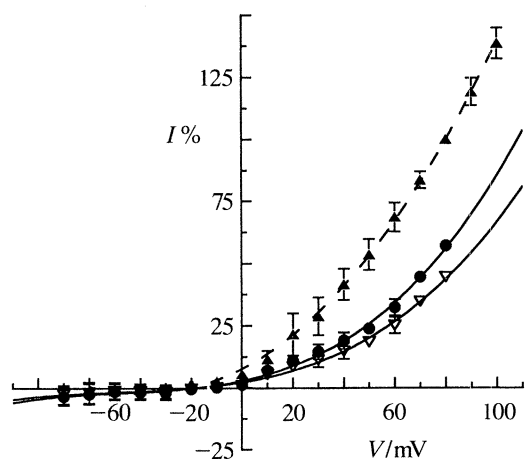


Figure 2. Effect of divalent cations on the  $\text{InsP}_3$ -induced current. The whole-vacuole current was measured either with  $\text{Ca}^{2+}$ ,  $\text{Ba}^{2+}$  or  $\text{Sr}^{2+}$  in the pipette and external solutions. The mean currents at different voltages were normalized to the mean current at  $+80$  mV with  $\text{Ca}^{2+}$  (410 pA), to reduce variability between vacuoles.  $I\%$  was found by multiplying the normalized values by 100. Symbols: filled triangles,  $\text{Ca}^{2+}$  ( $n=8$ ); filled circles,  $\text{Ba}^{2+}$  ( $n=3$ ); open triangles,  $\text{Sr}^{2+}$  ( $n=3$ ). Pipette medium:  $5 \times 10^{-3}$  M ( $\text{Ca}^{2+}$ ,  $\text{Ba}^{2+}$  or  $\text{Sr}^{2+}$ ),  $5 \times 10^{-3}$  M MES, pH adjusted to 5.5 with Tris. External medium:  $10^{-3}$  M ( $\text{Ca}^{2+}$ ,  $\text{Ba}^{2+}$  or  $\text{Sr}^{2+}$ ),  $5 \times 10^{-3}$  M Tris, pH adjusted to 7.5 with MES.

The voltage dependence of the channel for  $\text{Ba}^{2+}$  and  $\text{Sr}^{2+}$  was different from the one obtained with  $\text{Ca}^{2+}$ .

In animal cells, the glycosaminoglycan heparin has been reported to be an inhibitor of the  $\text{InsP}_3$ -induced  $\text{Ca}^{2+}$  current. In plants, Brosnan & Sanders (1990) report an inhibition of this current in red beet microsomes, by  $5 \mu\text{g ml}^{-1}$  heparin. In contrast, heparin has no effect on the  $\text{InsP}_3$ -mediated  $\text{Ca}^{2+}$  release of *Neurospora crassa* vacuoles (Cornelius *et al.* 1989). In a set of experiments with red beet vacuoles, we have examined the effect of heparin on the  $\text{InsP}_3$ -induced current in the whole-vacuole configuration, with high extracellular  $\text{Ca}^{2+}$  concentration ( $10^{-3}$  M). A 60% decrease in the  $\text{InsP}_3$ -induced current was obtained (figure 3) with  $1 \text{ mg ml}^{-1}$  heparin (CHOAY, S. A, Paris, France). To verify if heparin was effective specifically on the  $\text{InsP}_3$ -induced current, we looked for a possible effect of heparin on another

current in the tonoplast. The tonoplast of red beet vacuoles contains a  $\text{Ca}^{2+}$  sensitive current. In  $10^{-3}$  M  $\text{Ca}^{2+}$ , this current exhibits a strong rectification (figure 4, open diamonds), but remains small for positive values of the vacuolar potential. In  $10^{-7}$  M free  $\text{Ca}^{2+}$  (figure 4, filled triangles), the current reaches values similar in size with the  $\text{InsP}_3$ -induced current. When vacuoles were clamped at  $+80$  mV in a low  $\text{Ca}^{2+}$  solution, the addition of  $1 \text{ mg ml}^{-1}$  heparin in the external medium reduced the current (figure 4) by  $46 \pm 4\%$  ( $n=4$ ). These experiments suggest that heparin decreases in a non specific way the  $\text{InsP}_3$ -induced current. Brosnan & Sanders find that heparin, a widely used inhibitor of  $\text{InsP}_3$  receptors in animal cells is effective on *Beta vulgaris*. The experiments that we report here show that if heparin inhibits the  $\text{InsP}_3$  induced current of red beet vacuole, it also inhibits another type of channel of this same vacuole. So even if heparin has already been used to

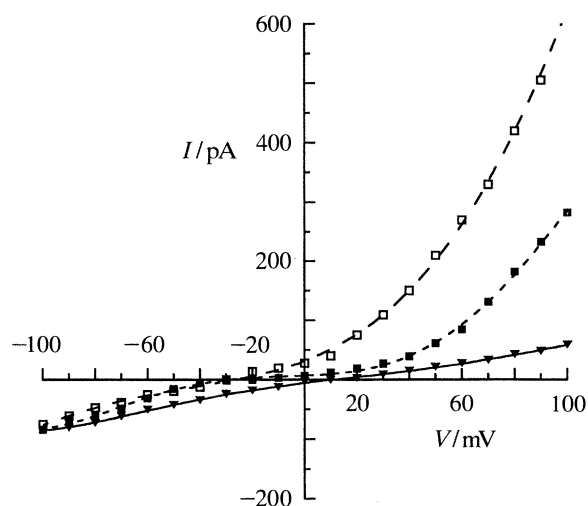


Figure 3. Effect of heparin on the  $\text{InsP}_3$ -induced current. The whole vacuole currents were measured on the same vacuole.  $I$ - $V$  curves were obtained, first without (filled triangles), then with  $10^{-6}$  M  $\text{InsP}_3$  (open squares) and finally when  $10^{-6}$  M  $\text{InsP}_3$  and  $1 \text{ mg ml}^{-1}$  heparin were added in the external medium (filled squares). Pipette medium:  $5 \times 10^{-3}$  M  $\text{CaCl}_2$ ,  $5 \times 10^{-3}$  M MES, pH adjusted to 5.5 with Tris. External medium:  $10^{-3}$  M  $\text{CaCl}_2$ ,  $10^{-4}$  M  $\text{ZnCl}_2$ ,  $5 \times 10^{-3}$  M Tris, pH adjusted to 7.5 with MES.

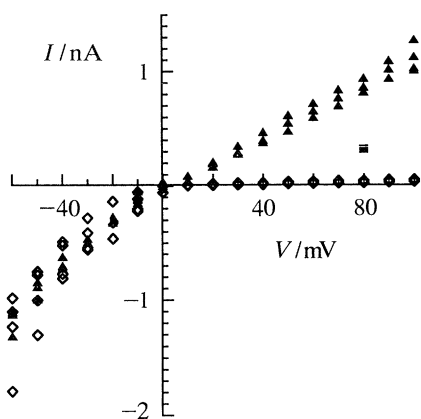


Figure 4. Effect of heparin on the Ca<sup>2+</sup>-sensitive current. *I-V* curves for four vacuoles were obtained with 10<sup>-3</sup> M CaCl<sub>2</sub> in the external medium (open diamonds). Another set of four vacuoles was used in the next experiments. For each of these vacuoles, an *I-V* curve (filled triangles) was obtained with 10<sup>-7</sup> M free Ca<sup>2+</sup> (1.5 × 10<sup>-3</sup> M Ca<sup>2+</sup> and 2 × 10<sup>-3</sup> M EGTA at pH 7.5) in the external medium. The vacuole was then clamped at +80 mV and 1 mg/ml<sup>-1</sup> heparin added in the external medium. A new steady-state value for the current was recorded (dashed line). Pipette medium: 5 × 10<sup>-3</sup> M CaCl<sub>2</sub>, 5 × 10<sup>-3</sup> M MES, pH adjusted to 5.5 with Tris. External medium: 5 × 10<sup>-3</sup> M Tris, pH adjusted to 7.5 with MES.

purify an InsP<sub>3</sub> receptor in rat cerebellum (Supattapone *et al.* 1988), it is doubtful that it can be used for purification of an InsP<sub>3</sub> receptor in plant cell membrane. Inhibition of the InsP<sub>3</sub>-induced Ca<sup>2+</sup> release by TMB8 also seems to require fairly large amounts of inhibitor (table 1).

#### (e) Properties of the Ca<sup>2+</sup> release by InsP<sub>3</sub>

The half-maximum InsP<sub>3</sub>-induced calcium release was obtained for an InsP<sub>3</sub> concentration, *K<sub>m</sub>*, varying between 0.2 and 5 μM on the different membranes studied (table 1). As a comparison, values between 0.1 and 1 μM are given for intracellular membranes in animal cells (Berridge & Irvine 1984; Ehrlich & Watras 1988; Watras *et al.* 1991). A cooperative opening of a Ca<sup>2+</sup> channel by InsP<sub>3</sub> has been mentioned in rat basophilic leukemia cells (Meyer *et*

*al.* 1988). However this is still controversial (Watras *et al.* 1991); in the experiments with plants listed (table 1), there is no report of a cooperative effect of InsP<sub>3</sub> on calcium release.

There is no mention of an inhibitory effect of high Ca<sup>2+</sup> concentration on the InsP<sub>3</sub>-induced current in plants. In our experiments on red beet vacuole, InsP<sub>3</sub> was still effective with millimolar concentrations of Ca<sup>2+</sup> in the extracellular medium (Alexandre *et al.* 1990). In animal cells, the sensitivity of the InsP<sub>3</sub>-mediated Ca<sup>2+</sup> release to high Ca<sup>2+</sup> concentrations is now explained by the regulation of the InsP<sub>3</sub> receptor by a Ca<sup>2+</sup> binding protein, sometimes associated to this receptor. However the InsP<sub>3</sub> receptor itself should be Ca<sup>2+</sup> insensitive (Supattapone *et al.* 1988).

#### (f) InsP<sub>3</sub> and endogenous Ca<sup>2+</sup>

In all the experiments quoted above, the Ca<sup>2+</sup> released by InsP<sub>3</sub> was preloaded into the organelle under study. In several experiments, microinjection of InsP<sub>3</sub> in a plant cell produces a signal by mobilizing Ca<sup>2+</sup> from internal stores (vacuole or endoplasmic reticulum). Using caged InsP<sub>3</sub> on stomatal guard cells of *Commelina communis*, Gilroy *et al.* (1990) observe an increase in cytosolic Ca<sup>2+</sup> and the closure of stomata. Blatt *et al.* (1990) also performed experiments with caged InsP<sub>3</sub>, on guard cells of *Vicia faba*. The effect of InsP<sub>3</sub> is detected by a change in activity of K<sup>+</sup> channels in the plasma membrane. In experiments by Thiel *et al.* (1990) and Förster (1990), microinjections of free InsP<sub>3</sub> in algal cells results in a change of the plasma membrane conductance. As previously noticed by Irvine (1990), from what is now known on the effect of InsP<sub>3</sub> in plant cells, it can be concluded that plants use InsP<sub>3</sub>, at least for the propagation of fast signals.

### 3. CALCIUM CHANGE AT THE SUBCELLULAR LEVEL

#### (a) Localized Ca<sup>2+</sup> change in the cytosol

The cascade of events following the opening of a Ca<sup>2+</sup> channel by InsP<sub>3</sub> starts by a change in cytosolic Ca<sup>2+</sup>, already noticed by several authors (Blatt *et al.* 1990;

Table 1.

material	authors	techniques	<i>K<sub>m</sub></i>	inhibitor
vacuolar membrane vesicles of Oat Roots	Schumaker & Sze (1987)	<sup>45</sup> Ca <sup>2+</sup>	0.6 μM	TMB8 250 μM
vacuoles <i>Neurospora crassa</i>	Cornelius <i>et al.</i> (1989)	<sup>45</sup> Ca <sup>2+</sup> Fura 2	5.28 μM	dantrolene <sup>a</sup>
vacuoles <i>Acer pseudoplatanus</i>	Ranjeva <i>et al.</i> (1988)	Quin 2	0.2 μM	TMB8 200 μM
microsomes <i>Beta vulgaris</i>	Brosnan & Sanders (1990, 1992)	<sup>45</sup> Ca <sup>2+</sup>	0.54 μM	TMB8 200 μM
vacuoles <i>Beta vulgaris</i>	Alexandre <i>et al.</i> (1990)	patch-clamp	0.22 μM	TMB8 100 μM
unicellular green alga <i>Eremosphaera viridis</i>	Förster (1990)	microelectrode		TMB8 200 μM

<sup>a</sup>There is no indication of the concentration used in this experiment

Gilroy *et al.* 1990; Thiel *et al.* 1990; Förster 1990). Opening an  $\text{InsP}_3$ -mediated  $\text{Ca}^{2+}$  channel is not the only way to modify cytosolic  $\text{Ca}^{2+}$ , and other  $\text{Ca}^{2+}$  channels and  $\text{Ca}^{2+}$  pumps have already been mentioned in plants (Schroeder & Thuleau 1991; Johannes *et al.* 1991). Modification of the distribution of cytosolic  $\text{Ca}^{2+}$  in space at a subcellular level has been reported in several experiments. Brownlee & Wood (1986), by use of microelectrode measurements, found higher values for cytoplasmic  $\text{Ca}^{2+}$  in rhizoid tip, than in sub-tip regions of *Fucus serratus*. Speknejder *et al.* (1989) studied the effect of various BAPTA-type buffers on the development of fucoid eggs. The building of a high calcium region located in the nascent tip seems to be an important step in the development. The authors found that millimolar amounts of the mobile BAPTA buffers seem to prevent a calcium increase in the nascent zone, by facilitating diffusion of  $\text{Ca}^{2+}$  away from the zone. Gilroy *et al.* 1991 used  $\text{Ca}^{2+}$ -sensitive indicators and image analysis to detect modification of cytosolic  $\text{Ca}^{2+}$  on guard cell of *Commelina communis*. Their experiments indicate that spatially localized increases in cytosolic  $\text{Ca}^{2+}$  occur around the nucleus and the vacuole, and precede stomatal closure.

The calcium signal consisting of spatial gradients and transient changes in  $\text{Ca}^{2+}$ , is very sensitive to the buffer strength in the cytosol. Blatt *et al.* (1990), showed that exogenous  $\text{InsP}_3$  reversibly inactivates  $\text{K}^+$  channels of the plasmalemma membrane of guard cells. Control is abolished by raising the  $\text{Ca}^{2+}$  buffer capacity of the cytoplasm with 20 mM EGTA or BAPTA. A similar effect has been obtained in characean algae (Thiel *et al.* 1990):  $\text{InsP}_3$  is not effective when added with 30 mM EGTA. In experiments on *Eremosphaera*, Förster (1990) used 20 mM BAPTA to abolish the activating effect of the  $\text{InsP}_3$ -mediated  $\text{Ca}^{2+}$  release on a  $\text{Ca}^{2+}$ -dependent  $\text{K}^+$  channel. In fucoid eggs, Speknejder *et al.* (1989) used 5 mM BAPTA to damp a spatial gradient.

#### (b) $\text{Ca}^{2+}$ diffusion in an homogeneous buffered medium

The importance of  $\text{Ca}^{2+}$  buffers on the propagation of a  $\text{Ca}^{2+}$  signal in the cytoplasm has been modelled by several authors under various experimental conditions. In the models, a change in cytosolic  $\text{Ca}^{2+}$ ,  $\delta\text{Ca}$ , initially occurs through opening of calcium channels, or activation of calcium pumps located in a membrane.  $\delta\text{Ca}$  is then propagated by diffusion in an isotropic medium containing various calcium buffers. The classical equations of simultaneous diffusion and chemical reactions are used (Crank 1967) to study the influence of the buffer parameters on the space and time dependence of  $\delta\text{Ca}$ . Simon & Llinas (1985) model the effect of a  $\text{Ca}^{2+}$  channel opening in a medium containing a mobile buffer. The authors indicate that the change in cytoplasmic  $\text{Ca}^{2+}$  is highly localized and may be effective only immediately near the channel itself. The perturbation around a cardiac  $\text{Ca}^{2+}$  channel has been studied by Bers & Peskoff (1991). In their model, the  $\text{Ca}^{2+}$  buffer consists of the

fixed  $\text{Ca}^{2+}$  binding sites of the sarcolemma. The number of sites is about 0.6 site per micrometer, and the binding reaction between calcium ions and the membrane is linear. Under these conditions, the authors demonstrate that the  $\text{Ca}^{2+}$  binding sites may contribute significantly to the buffering of intracellular  $\text{Ca}^{2+}$ . However other intracellular  $\text{Ca}^{2+}$  buffers may also modify the  $\text{Ca}^{2+}$  spread near a channel mouth. Nunogaki & Kasai (1988) propose a model to describe the effect of a mobile buffer on the diffusion of protons near a  $\text{H}^+$  channel. The mobile buffer concentration is regarded as constant with respect to time and uniform with respect to space, because of the large buffer concentration compared to the free  $\text{H}^+$  concentration. The authors discuss the steady-state profile for  $\text{H}^+$  ions near a channel or in small vesicles. The model can be applied to the opening of  $\text{Ca}^{2+}$  channels in an intracellular compartment, because of the same basic assumption concerning the diffusion of  $\text{H}^+$  or  $\text{Ca}^{2+}$  ions. The control of cytosolic  $\text{Ca}^{2+}$  by perfusing a patch pipette with a mobile buffer was studied by Neher (1986) and Mathias *et al.* (1990). The mobile buffer may be considered as a shuttle, which carries  $\text{Ca}^{2+}$  from the pipette to the cell, then diffuses back when empty. The fixed buffers in the cytosol greatly increase the time for equilibration of the pipette and intracellular  $\text{Ca}^{2+}$ . In a more complicated model, Sala & Cruz (1990) consider a spherical cell in which three different  $\text{Ca}^{2+}$  buffers are present. A buffer is characterized by its capacity, mobility and kinetic constants. The  $\text{Ca}^{2+}$  signal at the cell surface incorporates pump, leak and  $\text{Ca}^{2+}$  channel. Up to 15 parameters are used in the simulations, to calculate the time dependency of cytosolic  $\text{Ca}^{2+}$  at several locations in the cell. As underlined by the authors, an adequate combination of the parameters determines the depth reached by the  $\text{Ca}^{2+}$  signal.

There is a general constraint in a theoretical model: it is difficult to make predictions when the parameters for the model are not determined with enough accuracy. This is the case with the models presented here because the values of the parameters of all the  $\text{Ca}^{2+}$  buffers in the cytosol have not yet been determined.

#### (c) Direct coupling between a $\text{Ca}^{2+}$ signal and a $\text{Ca}^{2+}$ -dependent channel

We will now present experimental data obtained on a system in which a  $\text{Ca}^{2+}$  signal is propagated by diffusion between a  $\text{Ca}^{2+}$  channel and a  $\text{Ca}^{2+}$ -sensitive channel.

In a previous paper, we reported indirect evidence for a  $\text{Ca}^{2+}$  change located near the tonoplast membrane of isolated red beet vacuoles (Alexandre & Lassalles 1990). The tonoplast contains several types of channels (Alexandre & Lassalles 1991): among them, an  $\text{InsP}_3$ -mediated  $\text{Ca}^{2+}$  channel and a  $\text{Ca}^{2+}$ -sensitive channel. A coupling by diffusion between the two channels was already suggested. To test this hypothesis, new experiments were performed in  $10^{-7}$  M free  $\text{Ca}^{2+}$  solution. Isolated vacuoles were clamped at +30 mV in different concentrations of

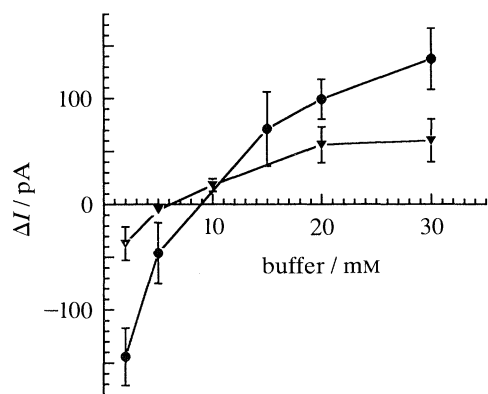


Figure 5. Effect of calcium buffers on the vacuolar current. Vacuoles were clamped at +30 mV and  $10^{-6}$  M  $\text{InsP}_3$  was added to the external medium. The change in vacuolar current  $\Delta I$ , was measured for different  $\text{Ca}^{2+}$  buffer strength in the external medium. The ratio  $r = \text{total Ca}^{2+} / (\text{total buffer} - \text{total Ca}^{2+})$  was kept constant:  $r=3$  for EGTA (filled circles);  $r=1$  for BAPTA (filled triangles). These values correspond to a free  $\text{Ca}^{2+}$  concentration close to  $10^{-7}$  M for all the solutions used. The apparent dissociation constant for EGTA at pH 7.5 is:  $2.6 \times 10^{-8}$  M (Ringbom 1967), and for BAPTA:  $10^{-7}$  M (Tsien 1980). Pipette medium:  $5 \times 10^{-3}$  M  $\text{CaCl}_2$ ,  $5 \times 10^{-2}$  M MES, pH adjusted to 5.5 with Tris. External medium:  $5 \times 10^{-2}$  M Tris, pH adjusted to 7.5 with MES. Filled circles and triangles represent mean values ( $n=4$ )  $\pm$  s.d.

EGTA and BAPTA buffers. The change in vacuolar current induced by  $10^{-6}$  M  $\text{InsP}_3$  was measured as follows, for each selected buffer solution. The steady state value  $I_1$  was first recorded. When  $\text{InsP}_3$  was added to the extra-vacuolar medium, the vacuolar current reached a new value,  $I_2$ . Variations of  $\Delta I = I_2 - I_1$  as a function of buffer concentrations are represented in figure 5. In BAPTA experiments,  $\Delta I$  reached a plateau for the buffer concentrations larger than 20 mM. For the smaller concentrations and for all the EGTA concentrations used,  $\Delta I$  strongly depended on the buffer concentration. A plateau is expected for the large values of the  $\text{Ca}^{2+}$  buffer preventing coupling by diffusion between the channels. Under these conditions, the  $\text{Ca}^{2+}$ -sensitive current and the  $\text{InsP}_3$ -induced  $\text{Ca}^{2+}$  current will not be affected by a change in buffer concentrations and  $\Delta I$  will remain constant. When the buffer concentration is decreased to the point where  $\text{Ca}^{2+}$  released by  $\text{InsP}_3$  start closing  $\text{Ca}^{2+}$ -sensitive channels,  $\Delta I$  will decrease. Interpretation of  $\Delta I$  for EGTA is less obvious because the  $\text{H}^+$  released by EGTA can modify the  $\text{Ca}^{2+}$ -sensitive current and the affinity of  $\text{InsP}_3$  for membrane receptors (Worley *et al.* 1987).

$\text{InsP}_3$  is not the only parameter controlling a  $\text{Ca}^{2+}$  channel in the tonoplast. Johannes *et al.* (1992) report that a voltage gated  $\text{Ca}^{2+}$  channel is present in the sugar beet vacuole. More generally, other pathways allowing a  $\text{Ca}^{2+}$  exchange between the vacuole and the cytoplasm are probably still unknown. However, our experiments with the  $\text{InsP}_3$ -gated  $\text{Ca}^{2+}$  channel suggest that, whenever one of the  $\text{Ca}^{2+}$  fluxes through the tonoplast is modified, the  $\text{Ca}^{2+}$ -sensitive channel can be affected.

A coupling by diffusion between channels by  $\text{Ca}^{2+}$  has already been mentioned by Marty & Neher (1985) in bovine chromaffin cells. The cell membrane contains a type of  $\text{K}^+$  channel which is activated by micromolar cytoplasmic  $\text{Ca}^{2+}$ . According to authors, the capacity of the buffer (EGTA or BAPTA) solutions used, together with the values of the  $\text{Ca}^{2+}$  flux through the cell membrane, dismiss the possibility of an homogeneous increase of  $\text{Ca}^{2+}$  over the entire cytosolic compartment. Measurements with various buffer solutions strongly suggest the possibility of a local increase in  $\text{Ca}^{2+}$  near the  $\text{K}^+$  channels. 5.5 millimolar BAPTA nearly suppresses this effect. BAPTA is more efficient than EGTA, but there is no correlation between the activation of the  $\text{K}^+$  channel and the value of the equilibrium capacity of the buffer.

#### (d) Use of characteristic time and characteristic length in $\text{Ca}^{2+}$ diffusion problems

In the last part of this paper, we will try to understand why large amounts of  $\text{Ca}^{2+}$  buffers are necessary to suppress coupling by diffusion in the above experiments. We will use an approach inspired from Nunogaki & Kasai (1988), to approximate the effect of buffers on  $\text{Ca}^{2+}$  diffusion. We consider a medium in which the  $\text{Ca}^{2+}$  buffer properties can be lumped in a single equation:



$K_b$ ,  $K_d$ : rate constants

$K = K_d/K_b$ : dissociation constant of the buffer

$\text{Ca}$ ,  $\text{B}$ : equilibrium concentrations of free  $\text{Ca}^{2+}$  and buffer

$\text{B-Ca}$ : equilibrium concentrations of  $\text{Ca}^{2+}$ -bound buffer

Analysis by Crank (1967, p. 121) was first used to solve the problem for the diffusion of  $\text{Ca}^{2+}$  in a buffered medium, by assuming that the buffering reaction (1) occurs instantaneously and that the buffer is never saturated. Under these conditions,  $\text{Ca}^{2+}$  appear to be diffusing in a buffered medium with an apparent diffusion constant,  $D'$ , smaller than the diffusion constant,  $D$ , of calcium ions in a free medium. However this approach has been questioned because buffering is diffusion limited and buffer is never of infinite concentration (Simon & Llinas 1985, Irving *et al.* 1990).

Another simplified approach to  $\text{Ca}^{2+}$  diffusion in a buffered medium is possible if we consider only small perturbations of the free ionic concentration. If  $\text{B}$  and  $\text{B-Ca}$  are much larger than  $\text{Ca}$ , they can be regarded as constant with respect to time and uniform with respect to space (Nunogaki & Kasai 1988). From reaction (1), a homogeneous perturbation,  $\delta\text{Ca}$ , at time  $t_0$  will be damped by the buffer with a time constant  $t_b$ ,

$$t_b = \frac{1}{K_b \text{B}}. \quad (2)$$

Equation 2 can be considered as a measurement of the temporal buffering power of the solution for small perturbations.

In the following we will consider that for the time smaller than  $t_b$ , the buffer has not enough time to react, so  $\text{Ca}^{2+}$  can freely diffuse over a distance  $r_b$ . In many problems of diffusion in homogeneous media,  $r_b$  can be approximated by:

$$r_b \approx 2\sqrt{Dt_b}. \quad (3)$$

Equation (3) gives approximatively (i) the distance,  $r_b$ , reached by a perturbation (initially located at  $r=0$ ) after a time  $t_b$ , and (ii) the time  $t_b$  necessary to damp a perturbation initially located in a spherical volume of radius  $r_b$  (Crank 1967).

From equations (2) and (3):

$$r_b \approx 2\sqrt{\frac{D}{K_b B}}. \quad (4)$$

Equation (4) gives an estimate of the spatial buffering power of a solution, e.g. the spatial range of a small perturbation in a buffered medium.

For numerical simulations in the cytosol:  $D = 6 \times 10^{-6} \text{ cm}^2 \text{ s}^{-1}$  (Bers & Peskoff 1991; Speksnijder *et al.* 1989) and  $K_b$  is in the range  $10^6$ – $10^8 \text{ M}^{-1} \text{ s}^{-1}$ , which corresponds to the difference between slow buffers like EGTA and fast buffers like BAPTA (Tsien 1980). Values for B in the cytosol are difficult to appreciate and estimations vary from  $\mu\text{M}$  to  $\text{mM}$ .

We can compare the values for  $r_b$ ,  $t_b$  (table 2) to the corresponding values of the perturbation brought by a channel opening in a non buffered medium. The distribution of ions around a channel mouth has already been described by Crank (1967) and Bers & Peskoff (1991). Calculations indicate the possibility of substantial elevation of  $\text{Ca}^{2+}$  at  $\mu\text{m}$  distance from the channel. We can also use equation (3) to estimate the perturbation brought by the channel. The mean time,  $t_o$ , during which the channel is open allows  $\text{Ca}^{2+}$  to diffuse a distance  $r_o$ . For the  $\text{InsP}_3$ -gated  $\text{Ca}^{2+}$  channel of the red beet vacuole (figure 1), and for the voltage-gated  $\text{Ca}^{2+}$  channel of sugar beet vacuole (Johannes *et al.* 1992),  $t_o$  is a few tenths of ms: this gives for  $r_o$  a value in the micrometer range. In most cases, table 2 indicates that  $r_b$  is smaller than  $r_o$ , so the buffer will damp the perturbation induced by the channel.

In the red beet vacuoles, the mean densities for the  $\text{Ca}^{2+}$ -sensitive (data not shown) and  $\text{Ca}^{2+}$ -selective channels (Alexandre *et al.* 1990) are about one channel per square micrometer. If the two-channel

Table 2. Values of  $r_b$ ,  $t_b$  for a fast buffer ( $r_{bf}$ ,  $t_{bf}$ ;  $K_b = 10^6 \text{ M}^{-1} \text{ s}^{-1}$ ) and for a slow buffer ( $r_{bs}$ ,  $t_{bs}$ ;  $K_b = 10^8 \text{ M}^{-1} \text{ s}^{-1}$ )

B	100 $\mu\text{M}$	1 mM	10 mM
$t_{bs}/\text{s}$	$10^{-2}$	$10^{-3}$	$10^{-4}$
$t_{bf}/\text{s}$	$10^{-4}$	$10^{-5}$	$10^{-6}$
$r_{bs}/\mu\text{m}$	5	1.5	0.5
$r_{bf}/\mu\text{m}$	0.5	0.15	0.05

populations are randomly distributed on the tonoplast, the mean distance,  $r_d$ , between channels will be in the micrometer range. There will be no coupling by diffusion between channels when  $r_b$  is much smaller than  $r_d$ . Data from table 2 may explain the lack of efficiency of EGTA buffers and the need of at least millimolar concentrations of BAPTA to prevent coupling in the experiments described figure 5.

#### (e) Importance of diffusion for $\text{Ca}^{2+}$ signals

The discussion on coupling by diffusion of  $\text{Ca}^{2+}$  presented in this paper, depends on the assumption that a  $\text{Ca}^{2+}$  perturbation never exceeds the capacity of the buffer. This is probably true (except very near the channel mouth) for *in vitro* experiments, in which large amounts of  $\text{Ca}^{2+}$  are used. It is more questionable *in vivo*, and we cannot predict exactly how far a  $\text{Ca}^{2+}$  signal can be propagated by diffusion in the cytosol. Regions with a fast buffer will reduce coupling to a distance much shorter than regions with a slow buffer. Variations in the buffer concentration also modify coupling and there are indications that concentration of a  $\text{Ca}^{2+}$  buffer like calmodulin may not be uniform in the cytosol (Marmé 1989). Differences in kinetic properties and concentrations of the buffer between various parts of the cytosol could then be used to regulate the  $\text{Ca}^{2+}$  signal propagation.

However it could be necessary to imply other processes than simple diffusion to explain  $\text{Ca}^{2+}$  changes in the cytoplasm. A rise in  $\text{Ca}^{2+}$  initially located in a small region can expand as a ' $\text{Ca}^{2+}$  wave' (Berridge & Irvine 1989). The wave could be propagated by diffusion on a short distance, triggering a  $\text{Ca}^{2+}$  release at a further point. A triggering mechanism, the  $\text{Ca}^{2+}$ -activated  $\text{Ca}^{2+}$  release, could then enhance the speed of the signal (Neylon *et al.* 1990). If such a trigger is sometimes mentioned in plant cells (Gilroy *et al.* 1990; Schroeder & Thuleau 1991), experimental evidence is still lacking to confirm this hypothesis.

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